

Home Search Collections Journals About Contact us My IOPscience

Electric field gradients in ¹¹¹In-doped $(Hf/Zr)_3AI_2$ and $(Hf/Zr)_4AI_3$ mixed compounds: *ab initio* calculations, perturbed angular correlation measurements and site preference

This article has been downloaded from IOPscience. Please scroll down to see the full text article. 2010 J. Phys.: Condens. Matter 22 215501 (http://iopscience.iop.org/0953-8984/22/21/215501)

View the table of contents for this issue, or go to the journal homepage for more

Download details: IP Address: 129.252.86.83 The article was downloaded on 30/05/2010 at 08:09

Please note that terms and conditions apply.

J. Phys.: Condens. Matter 22 (2010) 215501 (10pp)

Electric field gradients in ¹¹¹In-doped (Hf/Zr)₃Al₂ and (Hf/Zr)₄Al₃ mixed compounds: *ab initio* calculations, perturbed angular correlation measurements and site preference

L A Errico^{1,2}, H M Petrilli³, L A Terrazos⁴, A Kulińska^{5,6}, P Wodniecki⁵, K P Lieb⁶, M Uhrmacher⁶, J Belosevic-Cavor⁷ and V Koteski⁷

¹ Departamento de Física and Instituto de Física La Plata (IFLP, CONICET-UNLP), Facultad de Ciencias Exactas, Universidad Nacional de La Plata, CC 67, 1900 La Plata, Argentina

² Universidad Nacional del Noroeste Bonaerense (UNNOBA), Monteagudo 2772, Pergamino, CP 2700, Buenos Aires, Argentina

³ Instituto de Fisica, Universidade de Sao Paulo, CP 66318, 05315-970 Sao Paulo, SP, Brazil

⁴ Universidade Federal de Campina Grande, Centro de Educação e Saude, Cuite PB 5817-000, Brazil

⁵ Institute of Nuclear Physics Polish Academy of Sciences, Radzikowskiego 152,

31-342 Kraków, Poland

⁶ II. Physikalisches Institut, Georg-August-Universität Göttingen, Friedrich-Hund-Platz 1,

D-37077 Göttingen, Germany

⁷ VINCA Institute of Nuclear Sciences, 11001 Belgrade, Serbia

Received 23 December 2009, in final form 15 March 2010 Published 30 April 2010 Online at stacks.iop.org/JPhysCM/22/215501

Abstract

The quadrupolar hyperfine interactions of in-diffused ¹¹¹In \rightarrow ¹¹¹Cd probes in polycrystalline isostructural Zr₄Al₃ and Hf₄Al₃ samples containing small admixtures of the phases (Zr/Hf)₃Al₂ were investigated. A strong preference of ¹¹¹In solutes for the contaminant (Zr/Hf)₃Al₂ minority phases was observed. Detailed calculations of the electric field gradient (EFG) at the Cd nucleus using the full-potential augmented plane wave + local orbital formalism allowed us to assign the observed EFG fractions to the various lattice sites in the (Zr/Hf)₃Al₂ compounds and to understand the preferential site occupation of the minority phases by the ¹¹¹In atoms. The effects of the size of the supercell and relaxation around the oversized In and Cd probe atoms were investigated in detail.

1. Introduction

Hyperfine interaction techniques such as perturbed angular correlation (PAC) and Mössbauer spectroscopy can provide structural analysis of intermetallic compounds doped with very low concentrations of impurity (probe) atoms and often lead to definite conclusions on their lattice locations in polycrystalline materials [1–3]. In the last decade, a systematic study of the electric field gradients (EFG) of ¹⁸¹Hf/¹⁸¹Ta

and ¹¹¹In/¹¹¹Cd probe nuclei in the full series of hafnium and zirconium aluminides has been carried out, intermetallic compounds which exhibit a large range of rather simple lattice structures [3–11]. Recently, successful attempts have been made for the compounds Zr_4Al_3 and Hf_4Al_3 to assign the hyperfine interaction parameters of ¹⁸¹Hf/¹⁸¹Ta probes to the possible Hf and Zr lattice sites [2] and even to distinguish among the lattice structures deduced from x-ray diffraction [12]. The (Hf/Zr)_4Al_3 phase structure can be



Figure 1. X-ray diffraction spectra of a mixed $Hf_4Al_3-Hf_3Al_2$ sample after annealing at 1273 and 1373 K. The expected diffraction peaks of Hf_4Al_3 and Hf_3Al_2 according to the JCPDS database are marked.

described as P6/mmm (No 191) with one 3f (*mmm*) Wyckoff position of low symmetry for the Al atoms and two axially symmetric positions, labeled 2d (-6m2) and 2e (6mm), for the Zr/Hf atoms [4, 11]. Indeed, the two EFGs measured by means of PAC spectroscopy with ¹⁸¹Hf/¹⁸¹Ta probes have given evidence of substitutional Zr/Hf sites [2], as confirmed by EFG calculations using the full-potential augmented plane wave + local orbital formalism [13, 14]. As expected, in all the investigated hafnium and zirconium aluminides [4–9], the ¹⁸¹Hf/¹⁸¹Ta probes were found to occupy all the possible, nonequivalent Hf/Zr crystallographic sites.

The present work addresses the much more complicated problem of which lattice sites ¹¹¹In/¹¹¹Cd probes prefer in these aluminides. Evidently these probe atoms are different from the constituents of the compounds. Frequently, probe positions are tentatively attributed on the basis of either the EFG asymmetry parameter predicted by the point charge model, or by arguments of the atomic volumes and charges of solute and host atoms. However, such simplistic predictions often fail, as has been demonstrated for several Hf-Al and Zr-Al compounds. In the case of dilute 111 In $\rightarrow ^{111}$ Cd impurities in these aluminides it has been observed that some of the Zr/Hf sites may be preferred [9]. Indium and aluminum, belonging to the same group in the periodic table, and with their small difference in electronegativity clearly would favor substitution of Al atoms by the In probes. On the other hand, the atomic radii [15] favor oversized In impurities to substitute Hf or Zr instead of Al. This 'competition' between atomic size and electronegativity occurs in various Hf and Zr aluminides and, indeed in the case of ¹¹¹In-doped HfAl₂ [6], a reversible switching process between both types of lattice sites with temperature occurs.

A different type of site preference has been observed in a mixture of ZrAl and Zr₂Al₃ phases, where ¹¹¹In segregates in the trace Zr₂Al₃ phase [7]. Similar segregations of In solutes in two-phase mixtures have been found by Collins and Zacate [1] in the case of the Pd–Ga, Ni–Al and Fe–Al systems. Preference of a solute for a certain phase occurs when the difference in the Gibbs free energy ΔG is different from zero and is termed segregation to the phase with lower energy. Even small differences in enthalpy may produce a high degree of segregation.





Figure 2. Same as figure 1 for a mixed Zr_4Al_3 – Zr_3Al_2 sample after annealing at 1223 and 1293 K and compared with the JCPDS database standards of Zr_4Al_3 and Zr_3Al_2 .

In view of these uncertainties, we report here on the measurement and interpretation of the quadrupole interactions of ¹¹¹In \rightarrow ¹¹¹Cd impurities dissolved in (Zr/Hf)₄Al₃, in the presence of small (Zr/Hf)₃Al₂ phase admixtures. Preliminary results for the measured EFGs of ¹¹¹Cd in these mixed (Zr/Hf)₄Al₃ samples have been reported in [11]. Experimental results on the hyperfine interaction of implanted 111 In \rightarrow ¹¹¹Cd ions in (Zr/Hf)₃Al₂ have been communicated in [9]. Considering the occurrence of phase mixtures and a large number of possible lattice sites in each phase, it is only through first-principles EFG calculations of dilute ¹¹¹Cd atoms that unambiguous evidence for the lattice sites selected by the ¹¹¹In probes in these aluminides can be achieved. This has been recently demonstrated for the case of dilute Cd impurities in the Laves phase $(Hf/Zr)Al_2$ [10]. In the course of the present calculations, several effects have been considered influencing the EFGs, such as lattice distortions due to the oversized Cd probe atoms, the size of the supercell embedding the probes and the type of first-principles approximations.

2. Experimental details

Samples of Zr₄Al₃ (42.8 at.% Al) and Hf₄Al₃ (43.2 at.% Al) were produced by arc melting, under an argon atmosphere. During melting of the samples, they were doped with the carrier-free ¹¹¹In activity and annealed in evacuated and sealed quartz tubes at different temperatures. The nominal sample compositions were determined by the masses of the components. Small mass losses during arc melting were assumed to be caused by evaporation of the more volatile Al element. The Hf-Al and Zr-Al phase diagrams around 42.85 at.% Al concentration corresponding to (Hf/Zr)₄Al₃ stoichiometry are rather complicated [4], in particular for the Zr-Al system, and we did not succeed in synthesizing single-phase samples. Phase analyses done by means of xray powder diffraction (XRD) established the structure of the investigated samples being predominantly (Hf/Zr)₄Al₃ with some admixtures of the (Hf/Zr)₃Al₂ phases, whose intensities strongly decreased for increasing annealing temperature. The respective x-ray diffraction spectra are illustrated in figures 1 and 2 for annealing temperatures ranging up to 1373 K. Annealing of the Hf_4Al_3 sample was done at 1223 K (1 day)



Figure 3. PAC spectra -R(t) and their Fourier transforms $P(\omega)$ measured at 24 and 1073 K for ¹¹¹Cd in the Hf₄Al₃ sample annealed at 1373 K.



Figure 4. Temperature behavior of the various ¹¹¹Cd probe fractions in the Hf₄Al₃ sample. The 4(d), 4(f), 4(g) and 8(j) crystallographic sites relate to the Hf₃Al₂ minority phase.

and then at 1373 K (1 day), annealing of the Zr_4Al_3 sample at 1223 K (1 day) and afterwards at 1293 K (3 days).

The PAC experiments were carried out in the temperature range between 24 and 1100 K, those above 300 K using a small resistive oven of a low γ -ray absorption, and those below room temperature in a closed-cycle helium cryostat. A four BaF₂ detector apparatus which had a time resolution of 0.8 ns for the 171–245 keV cascade in ¹¹¹Cd was used.

For polycrystalline samples and a nuclear spin I = 5/2 of the intermediate nuclear level of the cascade, the perturbation factor $G_{22}(t)$ for static electric quadrupole interactions has the form [16]

$$G_{22}(t) = \sum_{i=1}^{k} f_i \sum_{n=0}^{3} s_{2n}(\eta_i) \cos(g_n(\eta_i)\nu_{Q_i}t) \exp(-g_n(\eta_i)\delta_i t)$$
(1)

where f_i is the relative fraction of nuclei that experience a given perturbation. No texture was present in the samples. The

various observed fractions f_i with different PAC parameters indicate the population of non-equivalent probe sites in the sample. The frequencies ω_n are related to the quadrupole frequency $v_Q = eQV_{33}/h$ by $\omega_n = g_n(\eta)v_Q$. The g_n and s_{2n} coefficients are known functions [17] of the asymmetry parameter $\eta = (V_{XX} - V_{YY})/V_{ZZ}$, where V_{ii} are the components of the diagonalized EFG tensor arbitrarily chosen as $|V_{XX}| \leq |V_{YY}| \leq |V_{ZZ}|$. The exponential functions account for a Lorentzian frequency distribution of relative width δ around ω_n . In these analyses the quadrupole moment value Q = 0.83(13)b for the isomeric ¹¹¹Cd PAC state [18] was adopted, but its error was not taken into consideration. Further details of the equipment used and data analysis are presented in [6, 16].

3. Experimental results

3.1. Hafnium aluminides

Figure 3 illustrates PAC spectra, R(t), and their Fourier transforms, $P(\omega)$, measured at 24 and 1073 K in the Hf₄Al₃ sample, which had been annealed for 1 day at 1373 K. The spectra are quite complex and exhibit four EFG fractions, one large fraction $f_1 = 60-80\%$ and three smaller fractions in the range of 5–20%. Their parameters are listed in table 1 and their assignments to lattice sites (in Hf₃Al₂) will be discussed in section 4. The evolutions of all four fractions f_i with the measuring temperature T_m are shown in figure 4; the fractions vary rather little with temperature.

In order to make the present results more transparent, we briefly refer to our previous analysis communicated in [11]. After 1 day of annealing the Hf₄Al₃ sample at 1223 K, the PAC spectrum taken at room temperature contained four EFG fractions, whose quadrupole parameters had been very similar to the ones previously found for ¹¹¹In/¹¹¹Cd probes in Hf₃Al₂ [9]. In the previously published analysis of the room-temperature data [11], the two largest fractions had been identified. Their quadrupole parameters were quoted as $v_{Q1} = 96.4(5)$ MHz and $\eta_1 = 0.69(1)$ for the largest fraction f_1 (possibly the Hf 4(g) site), and $v_{Q2} = 128(1)$ MHz and $\eta_2 = 0.44(1)$ for the second largest fraction f_2 (tentatively attributed to the Al 8(j) site). On the basis of these EFG



Figure 5. Same as figure 3 for the Zr_4Al_3 sample annealed at 1223 K.

Table 1. EFG parameters that characterize the hyperfine interactions (HFI) experimentally observed at ¹¹¹Cd probes in the mixed $(Hf/Zr)_4Al_3-(Hf/Zr)_3Al_2$ samples measured at room temperature and extrapolated to T = 0 K.

Sample	HFI	f (RT) (%)	v _Q (RT) (MHz)	η (RT)	$ V_{zz} $ (RT) (10 ²¹ V m ⁻²)	$ V_{zz} (0 \text{ K})$ (10 ²¹ V m ⁻²)
Zr ₄ Al ₃ -Zr ₃ Al ₂	1 2 3 4	21(3) 15(5) 23(2) 15(2)	115(2) 84(3) 34(2) 97.7(7)	0.21(5) 0.08(5) 0.30(3) 0.92(4)	5.73(5) 4.2(1) 1.69(5) 4.87(3)	5.9(1) 4.5(2) 1.7(1) 5.1(1)
Hf ₄ Al ₃ -Hf ₃ Al ₂	5 1 2 3 4	26(3) 13(5) 7(2) 63(3) 20(2)	130(1) 88(5) 39(2) 96.4(5) 128(1)	$\begin{array}{c} 0.30(4) \\ 0 \\ 0.12(5) \\ 0.69(3) \\ 0.44(2) \end{array}$	6.48(5) 4.4(3) 1.9(1) 4.80(3) 6.40(5)	6.9(1) 4.3(4) 2.0(2) 5.0(1) 6.7(1)

parameters, and especially the temperature evolution of the quadrupole frequencies, we had concluded that the ¹¹¹In impurities preferentially select sites in the minority phase Hf₃Al₂. On the other hand, it should be noted that—in agreement with the small Hf₃Al₂ phase admixture evidenced by XRD—the PAC spectra for ¹⁸¹Hf \rightarrow ¹⁸¹Ta probes (self-atoms) in Hf₄Al₃ [2] had not shown any of the quadrupole frequencies characteristic of the Hf₃Al₂ phase [9].

3.2. Zirconium aluminides

Similar results had been reported for the mixed Zr₄Al₃ sample [11]. After the annealing at 1223 K, the room-temperature perturbation spectrum featured, besides several other components, quadrupole parameters similar to those measured in Zr₃Al₂. Further annealing for 3 days at 1293 K resulted in an increase of the fraction with $\nu_{\rm Q} = 97.7(7)$ MHz and $\eta = 0.92(4)$ to about 50%, despite the decreasing Zr₃Al₂ phase admixture seen in the XRD spectrum (see figure 2).

The perturbation spectra and their Fourier transforms obtained at 24 and 1073 K measuring temperatures for this sample after the 1223 K annealing are presented in figure 5, while the temperature evolution of the fractions f_i (i = 1-5) is shown in figure 6. Again the assigned lattice sites will be discussed in section 4. Note that two of the fractions, labeled Al 8(j) and Al 3(f), vary quite strongly with the measuring temperature: the Al 3(f) fraction has disappeared around 600 K.

4

All the measured room-temperature EFG parameters are listed in table 1. Some of these parameters are slightly different from those communicated in the previous work [9, 11], due to the fact that in the new fits five instead of four EFG fractions were considered. The last column of table 1 lists the various EFGs extrapolated to T = 0, which are to be compared with the theoretical predictions for the various probe sites discussed below.

4. Theory

4.1. Procedure

The analysis of the PAC spectra for diffused ¹¹¹In \rightarrow ¹¹¹Cd impurities in the mixed (Hf/Zr)₄Al₃ + (Hf/Zr)₃Al₂ samples after annealing at the higher temperatures gave evidence for up to five fractions, each with different EFG parameters. By comparison with the results of previous PAC studies on ¹¹¹In-doped single-phase (Hf/Zr)₃Al₂ samples [9], four of the fractions could each be assigned to the (Hf/Zr)₃Al₂ minority phases in the mixed samples. The most important aim of the theoretical work was to assign the measured EFGs to the possible crystallographic sites by performing *ab initio* electronic structure calculations in all the relevant compounds. To this end, we determined the self-consistent potentials and charge densities inside the Cd-doped (Hf/Zr)₄Al₃ and (Hf/Zr)₃Al₂ hosts and then calculated the EFG tensors at a Cd probe nucleus replacing either a single Hf/Zr or Al atom in the

Table 2. Calculated and experimentally determined structural parameters for $(Zr/Hf)_4Al_3$ and $(Zr/Hf)_3Al_2$. Experimental data were taken from [12].

		Theory						
	Experiment	LDA	CW-GGA					
Zr_4Al_3								
а	5.433 Å	5.418 Å	5.430 Å					
С	5.390 Å	5.336 Å	5.394 Å					
x	0.250	0.260	0.259					
Hf_4Al_3								
а	5.343 Å	5.262 Å	5.343 Å					
С	5.422 Å	5.348 Å	5.476 Å					
x	0.250	0.260	0.258					
Zr ₃ Al ₂								
а	7.633 Å	7.661 Å	7.666 Å					
С	6.996 Å	7.001 Å	6.980 Å					
и	0.340	0.343	0.343					
v	0.200	0.199	0.199					
y	0.125	0.121	0.121					
Z	0.210	0.214	0.214					
	Hf ₃ Al ₂							
а	7.549 Å	7.571 Å	7.572 Å					
С	6.909 Å	6.906 Å	6.905 Å					
и	0.340	0.345	0.346					
v	0.200	0.202	0.202					
y	0.125	0.122	0.122					
z	0.210	0.211	0.211					

host lattice and taking properly into account the structural and electronic effects introduced by the impurity in the hosts. All the calculations refer to zero temperature and to substitutional probes.

As pointed out before the (Hf/Zr)₄Al₃ phase structure can be described as P6/mmm (no. 191), with one 3f (mmm) Wyckoff position of low symmetry for Al and two axially symmetric positions for the Zr/Hf atoms, labeled 2d (-6m2)and 2e(6mm) [4, 11]. The 2e atoms are located at (0, 0, x), with x = 1/4. The 3f and 2d are located at (1/2, 0, 1/2)0) and (1/3, 2/3, 1/2), respectively. $(Hf/Zr)_3Al_2$ crystallizes in the tetragonal tP20-type structure (space group $P4_2/mnm$ - D_{4b}^{14}), which possesses four non-equivalent crystallographic positions [4]: the Al atoms are in the 8(j) position of low symmetry, while the Hf/Zr atoms occupy either the axially symmetric 4(d) site or two sites of low m2m symmetry labeled 4(f) and 4(g). The positions of all the atoms in the unit cell are determined by the four internal parameters u, v, y and z, namely Hf/Zr 4(f) (u, u, 0), Hf/Zr 4(g) (v, 1 - v, 0) and Al 8(j)(y, y, z); the Hf/Zr 4(d) atoms are located at position (0, 1/2, 1/4). The experimental lattice parameters and the positional parameters u, v, y and z are listed in table 2.

In the case of $(Hf/Zr)_4Al_3$, the calculations were carried out considering a periodically repeated large supercell (SC), in which a Cd atom replaces a single host atom. Calculations were made for a periodic arrangement of 56-atom SCs, each constructed from eight unit cells of Hf₄Al₃ or Zr₄Al₃. The resulting SC has the dimensions a' = 2a = 10.866 Å,



Figure 6. Temperature dependence of the ¹¹¹Cd probe fractions in the Zr_4Al_3 sample. The 4(d), 4(f), 4(g) and 8(j) crystallographic sites concern the Zr_3Al_2 minority phase admixture. The rapidly decreasing fraction we tentatively attribute to the Al 3(f) site in the Zr_4Al_3 majority phase.

b' = 2b = 10.866 Å and c' = 2c = 10.788 Å (in the case of Zr₄Al₃) and a' = 2a = 10.686 Å, b' = 2b = 10.686 Å and c' = 2c = 10.844 Å (in the case of Hf₄Al₃) with a single host atom replaced by a Cd atom.

For the $(Hf/Zr)_3Al_2$ phase, since here the unit cell dimensions are larger than those of the (Hf/Zr)₄Al₃ phase, the calculations were performed for two different Cd concentrations. In this way, we can study the effect of Cd concentration on the electronic and structural properties of these doped systems. Firstly, a Cd atom substitutes a single Hf/Zr or Al atom in the 20-atom unit cell, i.e. ordered (Hf/Zr)₁₁CdAl₈ or (Hf/Zr)₁₂Al₇Cd alloys were considered. In a second approach, which approximates better (Hf/Zr)₃Al₂ with an *isolated* Cd impurity, we considered a periodically repeated larger supercell, in which a Cd atom replaces a single host atom. Calculations were made for a periodic arrangement of 160-atom supercells (SC), each constructed from eight unit cells of Hf₃Al₂ or Zr₃Al₂. The resulting SC has the dimensions a' = 2a = 15.266 Å, b' = 2b = 15.266 Å and c' = 2c =13.992 Å (in the case of Zr_3Al_2) and a' = 2a = 15.098 Å, b' = 2b = 15.098 Å and c' = 2c = 13.818 Å (in the case of Hf_3Al_2) with a single host atom replaced by a Cd atom.

In all cases, although the Cd concentration is large compared with the parts-per-million (ppm) dilution in the samples used in the PAC experiments, the choice of the described SCs keeps the Cd atoms sufficiently far from each other (more than 10 Å) to avoid significant impurity–impurity interactions.

To solve the scalar-relativistic Kohn–Sham equations, calculations based on the density-functional theory (DFT) were performed with the augmented plane wave plus local orbital (APW + lo) method [14, 19] as embodied in the WIEN2K code [13]. In this method the wavefunctions are expanded in spherical harmonics inside non-overlapping atomic spheres

of radius R_{MT} and in plane waves in the remaining space of the unit cell (the interstitial region). Exchange and correlation effects were treated within the density-functional theory using the local density approximation (LDA) [20] and the recently proposed general gradient approximation of Wu and Cohen (WC-GGA) [21]. For all calculations presented in this paper, the atomic spherical radii used for Cd, Zr, Hf and Al were 1.3 Å. The parameter Rk_{max} which controls the size of the basis set, was fixed to 8.0 for the pure systems and the smallest cells. In the case of the 160-atom supercells, we used $Rk_{max} =$ 7.0 (*R* is the smallest muffin tin radius and k_{max} the largest wavenumber of the basis set). Integration in the reciprocal space was performed using the tetrahedron method taking up to 5000 k-points in the first Brillouin zone for the pure systems and the smallest cells and 200 k-points for the 160-atom SCs. Once self-consistency of the potential was achieved, quantummechanically derived forces were obtained and the ions were displaced according to a Newton-damped scheme, and then the new positions for the atoms were obtained; for details, see [22, 23]. The procedure was repeated until the forces on the ions were below a tolerance value fixed as 0.05 eV \AA^{-1} ; in the case of the pure systems, the tolerance value was reduced to $2 \text{ meV } \text{\AA}^{-1}$. At each position the diagonal elements of the EFG tensor were obtained directly from the V_{2M} components of the lattice harmonic expansion of the self-consistent potential [24].

To check the accuracy of the results obtained in the 160atom SCs, we performed several additional calculations. For selected cases, we varied the basis set (number of plane waves) from $Rk_{max} = 5.0$ to 7.0. We also varied the number of *k*-points from 25 to 400. These variations showed that the EFGs and interatomic distances can be obtained with adequate precision using $Rk_{max} = 6.0$ and 200 *k*-points; for those values the interatomic distances and the EFG components converged within about 0.01 Å and 0.5×10^{21} V m⁻², respectively. Structural relaxation, leading to forces on the atoms less than 0.05 eV Å⁻¹, is the most important factor for convergence of the electronic properties, and they do not produce significant changes by increasing the number of plane waves or of *k*points.

4.2. Calculations for the undoped systems

In order to check the reliability of the theoretical approach, the equilibrium lattice parameters and positional parameters u, v, y and z in (Hf/Zr)₃Al₂ and x in (Hf/Zr)₄Al₃ were calculated. As the first step of this structural refinement, the atomic positions were relaxed. During this relaxation the lattice parameters and, as a consequence, the cell volume were kept fixed to their experimental values. Then the theoretical equilibrium volume was determined by fixing the atomic positions to their optimized values and further keeping the c/a ratio fixed. A series of calculations was carried out, in which the total energies were calculated as a function of the volume. In this way, we obtained the equilibrium volume (corresponding to the minimum energy). In the third step the c/a ratio was allowed to vary. Finally, the atomic positions were relaxed again, but now keeping the lattice parameters fixed at their optimized values. As can be seen

Table 3. Optimized distances d^{rel} (in Å) of the Cd impurity to its four first shells of neighbors in Cd-doped (Hf/Zr)₄Al₃ compared with the ones in the pure systems (d^{pure}), calculated principal EFG component, V_{ZZ} , in units of 10^{21} V m⁻², and asymmetry parameter η at the Cd impurity. In brackets the type and multiplicity of neighbors. Results correspond to the 56-atom supercells.

Site	d^{pure}	d^{rel}	V_{ZZ}	η
		Zr ₄ Al ₃		
Al 3f	2.66 Å [Al, 4] 3.00 Å [Zr, 4] 3.08 Å [Zr, 4] 4.61 Å [Al, 4] 4.76 Å [Zr, 4]	2.72 Å [Al, 4] 3.06 Å [Zr, 4] 3.12 Å [Zr, 4] 4.71 Å [Al, 4] 4.82 Å [Zr, 4]	+5.32	0.82
Zr 2d	3.07 Å [Zr, 3] 3.08 Å [Al, 6] 3.33 Å [Zr, 6] 4.86 Å [Al, 12]	3.15 Å [Al, 6] 3.18 Å [Zr, 3] 3.39 Å [Zr, 6] 4.86 Å [Al, 12]	+8.6	0
Zr 2e	2.55 Å [Zr, 1] 2.77 Å [Zr, 1] 3.00 Å [Al, 6] 3.30 Å [Zr, 6] 4.76 Å [Al, 6]	2.68 Å [Zr, 1] 2.81 Å [Zr, 1] 3.12 Å [Al, 6] 3.36 Å [Zr, 6] 4.95 Å [Al, 6]	-0.03	0
		Hf ₄ Al ₃		
Al 3f	2.63 Å [Al, 4] 2.97 Å [Hf, 4] 3.08 Å [Hf, 4] 4.56 Å [Al, 4]	2.68 Å [Al, 4] 3.03 Å [Hf, 4] 3.14 Å [Hf, 4] 4.63 Å [Al, 4]	+5.63	0.91
Hf 2d	3.04 Å [Hf, 3] 3.07 Å [Al, 6] 3.30 Å [Hf, 6] 4.83 Å [Al, 12]	3.10 Å [Hf, 3] 3.18 Å [Al, 6] 3.36 Å [Hf, 6] 4.90 Å [Al, 12]	+8.8	0
Hf 2e	2.57 Å [Hf, 1] 2.77 Å [Hf, 1] 2.97 Å [Al, 6] 3.30 Å [Hf, 6]	2.67 Å [Hf, 1] 2.82 Å [Hf, 1] 3.08 Å [Al, 6] 3.34 Å [Hf, 6]	-2.5	0

in table 2, the lattice constants and the parameters u, v, y, z and x were found in excellent agreement with the experimental data. Additionally, the results obtained using LDA and CW-GGA turned out to be very similar. For these reasons and in order to make the various calculations in the doped systems comparable, we fixed in the following the lattice parameters at the experimental values.

4.3. Calculations in the doped systems

After the study of the pure systems, we substituted an Hf/Zr or Al host atom by a Cd probe atom in the unit cells of $(Hf/Zr)_4Al_3$ and $(Hf/Zr)_3Al_2$, as described before. This substitution produces non-negligible forces on its nearest neighbors. We have to remark that even small changes in the atomic positions can induce large effects on the EFG and even more on the asymmetry parameter. For these reasons, full relaxation of all atomic positions were considered, until forces on the atoms were below 0.05 eV Å⁻¹. We use this tolerance criterion because a displacement induced by forces smaller than this limit produces changes in the EFG components that are below the convergence error. As in the case of the pure

Table 4. Same as table 3 for the pure and Cd-doped $(Hf/Zr)_3Al_2$ systems. The results were obtained at the relaxed structures of the 20-atom supercells. In brackets the type and multiplicity of neighbors.

		Ι	LDA		CW-GGA		
Site	d^{pure}	d^{rel}	V_{ZZ}	η	d^{rel}	V_{ZZ}	η
			Zr_3Al_2				
Al 8(j)	2.61 [Al] 2.78 [Zr] 2.83 [Zr] 2.93 [Zr, 2]	2.66 [Al] 2.84 [Zr] 2.87 [Zr] 2.95 [Zr, 2]	+5.1	0.08	2.65 [Al] 2.84 [Zr] 2.88 [Zr] 2.95 [Zr, 2]	+5.1	0.09
Zr 4(d)	3.05 [Al, 4] 3.26 [Zr, 4] 3.37 [Zr, 4] 3.50 [Zr, 2]	3.11 [Al, 4] 3.24 [Zr, 4] 3.38 [Zr, 4] 3.50 [Zr, 2]	-4.4	0	3.10 [Al, 4] 3.25 [Zr, 4] 3.38 [Zr, 4] 3.50 [Zr, 2]	-4.3	0
Zr 4(f)	2.83 [Al, 2] 2.93 [Al, 4] 3.37 [Zr, 4] 3.38 [Zr]	2.86 [A1, 2] 2.90 [A1, 4] 3.34 [Zr] 3.38 [Zr, 4]	+2.2	0.76	2.86 [Al, 2] 2.90 [Al, 4] 3.35 [Zr] 3.38 [Zr, 4]	+2.2	0.74
Zr 4(g)	2.78 [Al, 2] 2.93 [Al, 4] 3.26 [Zr, 4] 3.53 [Zr, 2]	2.70 [A1, 2] 3.01 [A1, 4] 3.24 [Zr, 4] 3.55 [Zr, 2]	-4.0	0.80	2.71 [Al, 2] 3.00 [Al, 4] 3.24 [Zr, 4] 3.54 [Zr, 2]	-3.9	0.80
			Hf_3Al_2				
Al 8(j)	2.60 [Al] 2.75 [Hf] 2.80 [Hf] 2.90 [Hf, 2]	2.63 [Al] 2.76 [Hf] 2.83 [Hf] 2.96 [Hf, 2]	+3.9	0.08	2.61 [Al] 2.75 [Hf] 2.82 [Hf] 2.95 [Hf, 2]	+3.8	0.11
Hf 4(d)	3.01 [Al, 4] 3.22 [Hf, 4] 3.34 [Hf, 4] 3.45 [Hf, 4]	3.06 [A1, 4] 3.22 [Hf, 4] 3.39 [Hf, 4] 3.45 [Hf, 4]	-4.2	0	3.06 [Al, 4] 3.21 [Hf, 4] 3.40 [Hf, 4] 3.45 [Hf, 4]	-4.2	0
Hf 4(f)	2.80 [Al, 2] 2.90 [Al, 4] 3.31 [Hf] 3.34 [Hf, 4]	2.83 [A1, 2] 2.85 [A1, 4] 3.29 [Hf] 3.38 [Hf, 4]	+1.2	0.37	2.82 [Al, 2] 2.85 [Al, 4] 3.28 [Hf] 3.38 [Hf, 4]	+1.3	0.44
Hf 4(g)	2.74 [Al, 2] 2.91 [Al, 4] 3.22 [Hf, 4] 3.49 [Hf, 2]	2.66 [A1, 2] 2.98 [A1, 4] 3.23 [Hf, 4] 3.52 [Hf, 2]	+3.8	0.97	2.67 [Al, 2] 2.98 [Al, 4] 3.21 [Hf, 4] 3.52 [Hf, 2]	+3.8	0.91

systems, both the LDA and CW-GGA were applied. Only small differences in V_{ZZ} of the order of 0.1×10^{21} V m⁻² and in the optimized distances (d^{rel}) by up to 0.01 Å were found between LDA and CW-GGA, while the asymmetry parameters were quite stable, providing differences between LDA and CW-GGA smaller than 0.02. Since the two approaches for the exchange and correlation potential gave differences of the order of the convergence errors, for simplicity only those obtained with LDA will be discussed here.

In order to find out which of the measured hyperfine interactions correspond to the majority phases $(Hf/Zr)_4Al_3$, and which to the minority phases $(Hf/Zr)_3Al_2$, the EFGs of Cd located at all possible crystallographic sites in all phases were calculated. We first present the results in the $(Hf/Zr)_4Al_3$ majority phases by showing in table 3 the relaxations of the Cd nearest neighbors (four shells of neighbors). In both Zr_4Al_3 and Hf_4Al_3 the structural distortions are quite anisotropic. As an example, in the case of the Zr 2d site the equilibrium Cd–Zr distance gets larger than the Cd–Al distance, in contrast to the unrelaxed structure.

The calculated EFG values for Cd at the three sites of $(Hf/Zr)_4Al_3$ are also listed in table 3. We first note that the calculations do not reproduce the measured EFGs (compare the experimental results of table 1 and APW + lo predictions in table 3). Regarding the typical accuracy of a few per cent for the calculated EFGs, no reliable site assignment is possible here. Only the EFG fraction observed in Zr₄Al₃ and characterized by $v_Q = 115$ MHz and $\eta = 0.21$ (HFI *bI*) may be tentatively assigned to the Al 3f site, although the predicted asymmetry parameter deviates from the experimental value.

In view of these negative results, we extended the calculations to the minority phases, $(Hf/Zr)_3Al_2$, by considering the two different Cd concentrations discussed before. The results obtained for the structural distortions as well as the values of V_{ZZ} and η for the 20-atom cells are reported in table 4. The Cd concentration in this case is far too high to allow a meaningful comparison with the PAC experiments for dilute systems. For this cell size, the interaction between Cd neighbors cannot be neglected and the structural relaxation of the Cd nearestneighbor atoms is constrained by the large Cd concentration

Table 5. Optimized distances d (in Å) of the Cd impurity to its four first shells of neighbors in Cd-doped (Hf/Zr)₃Al₂, calculated principal EFG component, V_{ZZ} , in units of 10^{21} V m⁻², and asymmetry parameter η at the Cd impurity. The theoretical results are compared with the experimental PAC results (extrapolated to 0 K). The results correspond to the relaxed structures of the 160-atom supercells.

		Theory			Experiment (0 K)		
System	Site	d	V_{ZZ}	η	V_{ZZ}	η	
Zr ₃ Al ₂	Al 8(j)	2.64 [Al] 2.85 [Zr] 2.85 [Zr] 2.89 [Zr, 2]	+6.9	0.26	6.9(1)	0.30(4)	
	Zr 4(d)	3.07 [Al, 4] 3.27 [Zr, 4] 3.40 [Zr, 4] 3.55 [Zr, 2]	-3.7	0.00	4.5(2)	0.08(5)	
	Zr 4(f)	2.84 [Al, 2] 2.94 [Al, 4] 3.27 [Zr] 3.37 [Zr, 4]	+1.5	0.34	1.7(1)	0.30(3)	
	Zr 4(g)	2.69 [Al, 2] 2.96 [Al, 4] 3.25 [Zr, 4] 3.56 [Zr, 2]	-4.6	0.85	5.1(1)	0.92(4)	
Hf ₃ Al ₂	Al 8(j)	2.62 [A1] 2.79 [Hf] 2.80 [Hf] 2.89 [Hf, 2]	+6.7	0.10	6.7(1)	0.44(2)	
	Hf 4(d)	3.03 [A1, 4] 3.24 [Hf, 4] 3.36 [Hf, 4] 3.50 [Hf, 2]	-3.9	0.00	4.3(2)	0.00	
	Hf 4(f)	2.81 [A1, 2] 2.90 [A1, 4] 3.24 [Hf] 3.41 [Hf, 4]	+1.1	0.17	2.0(1)	0.12(5)	
	Hf 4(g)	2.66 [A1, 2] 2.95 [A1, 4] 3.25 [Hf, 4] 3.54 [Hf, 2]	+4.6	0.62	5.0(1)	0.69(3)	

and the spurious interaction between relaxing atoms. Therefore calculations using the much larger SC were performed. Comparing the equilibrium Cd-bond lengths for the two cell sizes, it is seen in tables 4 and 5 that there are small differences, which exceed the convergence errors. In conclusion and not unexpectedly the structural distortion induced by the Cd probe depends on the impurity concentration. As the EFG generally decays as r^{-3} , where *r* is the distance from the producing charge density, it is particularly sensitive to slight local changes, especially in the first shells of neighbors. For this reason, the structural distortions must be carefully calculated when comparing the EFG predictions with the experimental results.

In table 5, the calculated EFG tensors at the Cd sites in the equilibrium structures of $(Hf/Zr)_3Al_2$ using the 160-atom SCs are summarized, which again illustrate the strong dependence on the Cd concentration. Since different concentrations produce different structural distortions, it is interesting to separate the electronic and structural effects. As a test case, we

8

Table 6. Calculated substitution energies of an In atom at the various sites in the $(Hf/Zr)_4Al_3$ and $(Hf/Zr)_3Al_2$. Results are obtained with the LDA approach and relaxed 160-atom supercell.

Phase	Site	Substitution energy (eV)
Hf ₄ Al ₃	Hf 2(d) Hf 2(e) Al 3(f)	1.8 1.7 2.7
Hf ₃ Al ₂	Hf 4(d) Hf 4(f) Hf 4(g) Al 8(j)	2.2 2.3 1.1 2.7
Zr ₄ Al ₃	Zr 2(d) Zr 2(e) Al 3(f)	5.8 5.9 2.5
Zr ₃ Al ₂	Zr 4(d) Zr 4(f) Zr 4(g) Al 8(j)	4.8 5.2 5.6 2.7

considered the 20-atom cell of Zr_3Al_2 , but fixed the Cd atom and its nearest neighbors at the equilibrium positions predicted for the 160-atom SC and calculated the EFG tensors for this geometry at all four possible Cd sites. As a result, for identical geometries of Cd, significant differences in the EFGs occur between the 20-atom and 160-atom SC. This calculation proves the relevance of the large Cd–Cd interaction in the case of the smaller cell, which affects the EFG even when identical ion positions are used, besides the effects of structural distortions based on different impurity concentration. Similar conclusions were drawn in the case of Hf_3Al_2 .

Finally, by comparing the measured EFG parameters with the ones obtained from APW + lo we can now assign the observed hyperfine interaction fractions to the various crystallographic sites in $(Hf/Zr)_3Al_2$. As illustrated in table 5, the theoretical predictions are in most cases in very good agreement with the experiments. However, deviations were found for the site Zr 4(d) in Zr₃Al₂ and the sites Al 8(j) and Hf 4(f) in Hf₃Al₂. In the first case, the discrepancy in V_{ZZ} amounts to about 20%, while in the last case, we find a serious discrepancy by a factor of two, probably due to the small value of V_{ZZ} at this site. For the purpose of this paper, we judge the few deviations as acceptable considering the complex mixed phases treated.

4.4. Phase preference of In dopants

If the various lattice sites in the mixed compounds were randomly occupied by the ¹¹¹In(\rightarrow ¹¹¹Cd) probes, the sites corresponding to the (Hf/Zr)₄Al₃ majority phases should be much more strongly populated. This is opposite to the experimental finding of the PAC measurements that the ¹¹¹In impurities, after the high temperature annealing, end up nearly exclusively on the sites in the (Hf/Zr)₃Al₂ minority phases. Note that, in order to explain this preferential site occupation, we have to consider the radioactive father of ¹¹¹Cd, namely ¹¹¹In, thus making the reasonable assumption that the ¹¹¹In \rightarrow ¹¹¹Cd electron capture decay does not change the lattice sites selected by indium.

Table 7. *Ab initio* based final assignments of the hyperfine interactions (HFI) experimentally observed at ¹¹¹Cd probes in the mixed $(Hf/Zr)_4Al_3-(Hf/Zr)_3Al_2$ samples to crystallographic sites.

Sample	HFI	f (RT) (%)	ν _Q (RT) (MHz)	η (RT)	$ V_{zz} $ (RT) (10 ²¹ V m ⁻²)	Site
Zr ₄ Al ₃ –Zr ₃ Al ₂	1 2 3 4 5	21(3) 15(5) 23(2) 15(2) 26(3)	115(2) 84(3) 34(2) 97.7(7) 130(1)	$\begin{array}{c} 0.21(5) \\ 0.08(5) \\ 0.30(3) \\ 0.92(4) \\ 0.30(4) \end{array}$	5.73(5) 4.2(1) 1.69(5) 4.87(3) 6.48(5)	Al 3f Zr 4(d) Zr 4(f) Zr 4(g) Al 8(j)
Hf ₄ Al ₃ -Hf ₃ Al ₂	1 2 3 4	13(5) 7(2) 63(3) 20(2)	88(5) 39(2) 96.4(5) 128(1)	0.00 0.12(5) 0.69(3) 0.44(2)	4.4(3) 1.9(1) 4.80(3) 6.40(5)	Hf 4(d) Hf 4(f) Hf 4(g) Al 8(j)

In order to understand the site selection, we performed *ab initio* total energy calculations for In-doped $(Hf/Zr)_4Al_3$ and $(Hf/Zr)_3Al_2$ compounds, in which In occupies the seven possible lattice locations in each mixed sample. The calculations were performed using the 56-atom SC in $(Hf/Zr)_3Al_2$ and the 160-atom Sc in $(Hf/Zr)_4Al_3$, with the same precision as that corresponding to the Cd impurities. The structural distortions induced by the In impurities in the host were also computed.

In order to estimate the preferred site occupation, we compared the defect formation energies, i.e. the substitution energies, E_{subst} , for both phases using the expression

$$E_{\text{subst}} = E_{\text{subst}}^{\text{tot}} - E_{\text{In}} + E_X - E_{\text{pure}}.$$
 (2)

Here $E_{\text{subst}}^{\text{tot}}$ is the total energy of the supercell containing a single substitutional In atom, E_{In} and E_X denote the energies of an In impurity and Hf/Zr or Al atom, and E_{pure} is the energy of the considered supercell without impurities. The substitution energies were obtained by calculating the energies of undoped (Hf/Zr)₄Al₃ and (Hf/Zr)₃Al₂ compounds, E_{pure} , and E_{In} and E_X from metallic In, Hf, Zr and Al with the same precision as that in the supercells. The convergence errors in the substitution energies are around 0.3 eV. In passing we note that small values of the substitution energies indicate preferential replacement of a host atom by indium, although the approach does not allow us to quantitatively estimate the hyperfine fractions.

The obtained defect formation energies for each site in all the compounds are reported in table 6. In the mixed $Hf_4Al_3-Hf_3Al_2$ sample, the substitution energy for the Hf 4(g) site in Hf_3Al_2 is the smallest one. Based on this argument, we may explain the preferential site occupation of this site observed in the PAC experiments (see table 1). In the case of Zr₄Al₃–Zr₃Al₂, the smallest substitution energies correspond to the Al 3(f) site in Zr_4Al_3 and the Al 8(j) site in Zr_3Al_2 . Again, when comparing the measured hyperfine interaction fractions with the ones expected on the basis of the total energy calculations, we can see that the interactions associated with Cd in the Al sites are those with larger populations, in agreement with our calculations for the substitution energies. The other interactions associated with the Zr 4(d), Zr 4(f) and Zr 4(g) have similar populations. Comparing to the calculated substitution energies, we see that the substitution energies for these sites are smaller than those for Cd substitutionally located in the Zr sites (Zr 2(d) and Zr 2(e)) of the Zr₄Al₃ phase.

In conclusion, the calculations of substitution energies make plausible the observed preference of In probes for the crystallographic sites in $(Hf/Zr)_3Al_2$. In light of the theoretical calculations performed here, the assignments of the hyperfine interactions observed in the PAC experiments can be done and are summarized in table 7.

5. Summary and conclusions

Hyperfine interactions and, in particular, electric field gradients constitute a very sensitive tool to investigate the local environment of probe nuclei; their measurement can provide detailed information on structural and electronic properties of the system under study. In a well-defined single-phase crystal with known lattice structure the interpretation of such measurements is usually not too difficult, but even here situations may occur, where distinguishable hyperfine fractions must be assigned to different lattice sites. In more complicated cases with impurities, point defects, multi-phase samples or systems with more than one crystallographic site, the interpretation of the various EFG fractions is by no means straightforward. Arguments based on point charge summations or symmetry considerations may fail, since the chemical nature of the impurity is not taken into account which, through structural distortions of its neighborhood, may strongly affect the EFG. To unravel these complex cases, a realistic theory is of great help, which models different structural and electronic scenarios.

The present work concerns such complex systems, namely coexistent Hf–Al intermetallic phases: samples of $(Hf/Zr)_4Al_3$ containing small amounts of $(Hf/Zr)_3Al_2$ and doped with low concentrations of the PAC probes ¹¹¹In \rightarrow ¹¹¹Cd. In principle, the impurity probe atoms can populate seven different crystallographic sites in these compounds. PAC revealed up to five EFG components in each sample. To assign these hyperfine interactions to the crystallographic sites, DFT calculations of electronic structure and atomic forces have been used to describe $(Hf/Zr)_4Al_3$ and $(Hf/Zr)_3Al_2$ in their pure forms as well as with substitutional Cd and In impurities.

In this work, only substitutional In and Cd probe impurities were considered, replacing an Hf/Zr or Al atom, and the supercell approach was employed. Although in theory the impurity concentration is considerably larger than in the experiment, the 56-atom and 160-atom supercell calculations used in the case of $(Hf/Zr)_4Al_3$ and $(Hf/Zr)_3Al_2$ yield results that sufficiently converged to allow comparison to the data. For all the systems, the defect structures were optimized using force calculations and from the self-consistent potentials the EFG were derived, allowing us a direct comparison to the experimental EFGs.

In the case of the pure systems, APW + lo correctly predicts the lattice parameters and internal positions of the Hf, Zr and Al atoms. Additionally, LDA and CW-GGA predict very similar equilibrium structures. Concerning the doped systems, the Cd impurities were found to induce sizable structural distortions in the host lattices. The relaxed distances were found almost independent of the approximations employed for the correlation and exchange potential. Due to the r^{-3} dependence of the EFG operator, the EFG is very sensitive to fine details of the atomic positions. Finally, we found that the calculations performed here (LDA, WC-GGA) predict almost the same EFGs for the equilibrium structures of Cd-doped Zr₄Al₃. This statement is also valid in the case of Hf₄Al₃ and the 3-2 phases (Cd-doped Zr₃Al₂ and Hf_3Al_2). Again, due to the fact that the EFG is very sensitive to small changes in the electronic charge density, we may conclude that these calculations give very similar descriptions of the electronic structures of each system.

The calculated substitution energies suggest that In prefers to locate at the Hf 4(g) sites in the Hf₃Al₂-Hf₄Al₃ sample. For this site, the substitution energy is 1.1 eV, which is at least 0.6 eV (which corresponds to at least 30%) smaller than the other ones. This fact explains the site preference of the ^{111}In \rightarrow ^{111}Cd probes (63% of the population observed in the experiments). The case of the Zr₃Al₂-Zr₄Al₃ sample is more subtle: the calculated substitution energies suggest that In prefers either the Al 8(j) or Al 3(f) sites, but these energies are very close together. In addition, the calculations for the Cd-doped compounds allowed us to assign a total of at least eight hyperfine interactions to the various inequivalent sites of these phases. These detailed calculations for solute In atoms in the $(Hf/Zr)_3Al_2$ and $(Hf/Zr)_4Al_3$ intermetallics, compared with TDPAC measurements, indicate that the In mother probes predominantly populate the sites in the minority phases (Hf/Zr)₃Al₂.

Acknowledgments

This work was partially supported by Agencia Nacional de Promoción Científica y Tecnológica (ANPCyT) (PICT98 03-03727), Consejo Nacional de Investigaciones Científicas y Técnicas (CONICET)(PIP6032 and 0002), Fund. Antorchas, Argentina, and Third World Academy of Sciences (TWAS), Italy (RGA 97-057), Cnpq, Fapesp and Capes (Brazil). This research made use of the HP-Parallel-Computing Bose Cluster and the computational facilities of the Physics of Impurities (PhI group) at IFLP and Departamento de Física (UNLP) and at LCCA-USP. We acknowledge support by Deutsche Forschungsgemeinschaft (DFG) and Deutscher Akademischer Austauschdienst (DAAD). JB-C and VK acknowledge the support from the Serbian Ministry of Science.

References

- Collins G and Zacate M 2001 Hyperfine Interact. 136/137 641 Collins G and Zacate M 2001 Hyperfine Interact. 136/137 647
- [2] Wodniecki P, Kulińska S, Wodniecka B, Cottenier S, Petrilli H M, Uhrmacher M and Lieb K P 2007 Europhys. Lett. 77 43001
- [3] Kulińska A, Wodniecki P, Wodniecka B, Petrilli H M, Terrazos L A, Uhrmacher M and Lieb K P 2009 J. Phys.: Condens. Matter 21 506001
- [4] Villars P and Calvert L D 1991 Pearson's Handbook of Crystallographic Data for Intermetallic Phases (Materials Park, OH: ASM)
 - Okamoto H 2000 Phase Diagrams for Binary Alloys (Materials Park, OH: ASM)
- [5] Wodniecki P, Wodniecka B, Kulińska A, Uhrmacher M and Lieb K P 2000 J. Alloys Compounds 312 17
 - Wodniecki P, Wodniecka B, Kulińska A, Uhrmacher M and Lieb K P 2001 *Hyperfine Interact.* **136/137** 535
- [6] Wodniecki P, Wodniecka B, Kulińska A, Uhrmacher M and Lieb K P 2001 Phys. Lett. A 288 227
- Wodniecki P, Wodniecka B, Kulińska A, Uhrmacher M and Lieb K P 2002 J. Alloys Compounds **335** 20
- [7] Wodniecki P, Wodniecka B, Kulińska A, Uhrmacher M and Lieb K P 2003 J. Alloys Compounds 351 1
- [8] Wodniecki P, Wodniecka B, Kulińska A, Uhrmacher M and Lieb K P 2004 J. Alloys Compounds 365 52
- [9] Wodniecki P, Kulińska A, Wodniecka B, Uhrmacher M and Lieb K P 2005 *Hyperfine Interact*. 158 429
 Wodniecki P, Kulińska A, Wodniecka B, Uhrmacher M and
- Lieb K P 2005 Hyperfine Interact. 158 339
 [10] Belosevic-Cavor J, Koteski V, Cekic B and Umecevic A 2007 Comput. Mater. Sci. 41 164
- [11] Wodniecki P, Kulińska A, Wodniecka B, Uhrmacher M and Lieb K P 2007 *Hyperfine Interact.* 177 111
- [12] Cenzual K, Gelato L M, Penzo M and Parthe E 1991 Acta Crystallogr. B 47 433
- [13] Blaha P, Schwarz K, Madsen G, Kvasnicka D and Luitz J 2001 WIEN2k, an Aug-Plane Wave + Local Orbitals Program for Calculating Crystal Properties Technische Universität Wien (ISBN 3-9501031-1-2)
- [14] Madsen G K H, Blaha P, Schwarz K, Sjöstedt E and Nordström L 2001 Phys. Rev. B 64 195134
- [15] Slater J C 1964 J. Chem. Phys. **39** 3199
- [16] Frauenfelder H and Steffen R 1963 Alfa-, Beta- and Gamma-Ray Spectroscopy vol 2, ed K Siegbahn (Amsterdam: North-Holland) p 917
- [17] Mendoza-Z élis L A, Bibiloni A G, Caracoche M C, López-García A R, Martínez J A, Mercader R C and Pasquevich A F 1977 Hyperfine Interact. 3 315
- [18] Herzog P, Freitag K, Reuschenbach M and Walitzki H 1980 Z. Phys. A 294 13
- [19] Andersen O K 1975 *Phys. Rev.* B 12 3060
 Sjöstedt E, Nordström L and Singh D J 2000 *Solid State Commun.* 114 15
 - See also Cottenier S 2002 Density Functional Theory and the Family of (L)APW-Methods: A Step-by-Step Introduction KU Leuven, Belgium http://www.wien2k.at/reg_user/ textbooks
- [20] Perdew J P and Wang Y 1992 Phys. Rev. B 45 13244
- [21] Wu Z and Cohen R E 2006 *Phys. Rev.* B **73** 235116
 [22] Errico L A, Fabricius G, Rentería M, de la Presa P and Forker M 2002 *Phys. Rev. Lett.* **89** 055503
- Errico L A, Fabricius G and Rentería M 2003 *Phys. Rev.* B 67 144104
- [23] Terrazos L A, Petrilli H M, Marszalek M, Saitovich H, Silva P R J, Blaha P and Schwarz K 2002 Solid. State Commun. 121 525
- [24] Schwarz K, Ambrosch-Draxl C and Blaha P 1992 Phys. Rev. B 42 2051
 - Schwarz K, Ambrosch-Draxl C and Blaha P 1992 *Phys. Rev.* B 46 1321